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#540

SECTION I NAME

24 HOUR EMERGENCY ASSISTANCE

Product	BURET SOLUTION A	CHEMTREC 800-424-8300	Health 3
Chemical Synonyms	Potassium Hydroxide Solution	Day 716-226-6177 Night 716-334-4222	Fire 0
Formula	KOH/Water Solution		Reactivity 2
Net(s) Size	100 ml. to 4 liters		
A.S. No.	1310-58-3		

SECTION II HAZARDOUS INGREDIENTS OF MIXTURES

Hazardous Component(s)	%	TLV Units
Potassium Hydroxide Water Solution	10%	2 mg/m ³ (air) (ACGIH 1984)

DANGER: ☠ POISON ☠ CAUSES SEVERE SKIN AND EYE BURNS

SECTION III PHYSICAL DATA

Melting Point (°F)	Freezes approx. 0°C	Specific Gravity (H ₂ O = 1)	Approx. 1.1
Boiling Point (°F)	Approx. 110-120°C	Percent Volatile by Volume (%)	90%
Vapor Pressure (mm Hg)	14 mm (water)	Evaporation Rate (Ether = 1)	Greater than 1
Relative Density (H ₂ O = 1)	0.7 (water)		
Solubility in Water	Complete		
Flammability and Odor	Clear, colorless liquid, no odor.		

SECTION IV FIRE AND EXPLOSION HAZARD DATA

Flash Point (°F)	Non-flammable liquid (NA)	Flammable Limits in Air (% by Volume)	NA	Lower	Upper
Ignition	If involved in a fire, flood with water, taking care not to splatter or splash.				

SECTION V SPECIAL FIREFIGHTING PROCEDURES

This material is highly corrosive. In fire conditions, wear a NIOSH-approved self-contained breathing apparatus and full protective clothing.

SECTION VI SPECIAL FIRE AND EXPLOSION HAZARDS

Will release flammable and explosive hydrogen gas when in contact with aluminum, lead, tin, zinc, and other alloys.

SECTION VII NON-REGULATED

Used by U.S. Department of Labor "essentially similar" to form OSHA-20

SECTION V HEALTH HAZARD DATA

Threshold Limited Value

TWA: Ceiling Limit Potassium Hydroxide 2 mg/m³ (air) (ACGIH, 1984) #1004

Effects of Overexposure

DANGER! CAUSES SEVERE BURNS. MAY BE FATAL IF SWALLOWED.
Avoid contact with skin, eyes and mucous membranes.

Emergency and First Aid Procedures

SKIN: Flood with water, then wash with vinegar.
EYES: Flush with water for 15 minutes. Get prompt medical attention.
INTERNAL: Drink several glasses of water. Follow with citrus fruit juice if available. Call a physician immediately.

SECTION VI REACTIVITY DATA

Stability	Unstable	Conditions to Avoid
Stable	x	Acids, excessive temperature and heat.
Incompatibility (Materials to avoid)		Common metals and their alloys; acids and their anhydrides; easily oxidizable compounds.

Hazardous Decomposition Products

None known.

Hazardous Polymerization

Will Not Occur

SECTION VII SPILL OR LEAK PROCEDURES

Steps to be taken in case material is released or spilled

Wear proper safety equipment. Dilute with water and neutralize with 6 Molar hydrochloric acid or sodium bisulfate.

Waste Disposal Method

Neutralize with sodium bisulfate and flush to sewer with copious amounts of water.

SECTION VIII SPECIAL PROTECTION INFORMATION

Respiratory Protection	None should be needed in normal laboratory handling. If necessary work in fume hood or wear a NIOSH-approved respirator.
Ventilation	Local Exhaust Recommended
Protective Gloves	Rubber
Eye Protection	Goggles or face shield.

SECTION IX SPECIAL PRECAUTIONS

Precautions to be Taken in Handling and Storing

Store in a cool place away from acids and acid fumes. Wash thoroughly after handling.

Other Precautions

Read label on container before using.

For laboratory use only. Not for drug, food or household use. Keep out of reach of children.

Rev. No. 1/12/87 Date 1/12/87 Approved [Signature] Chemical Safety Coordinator

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